

-1-

Angiogenesis inhibiting 5-substituted-1,2,4-thiadiazolyl derivatives

5 This invention concerns 5-substituted-1,2,4-thiadiazolyl derivatives acting as angiogenesis inhibitors, and their preparation; it further relates to compositions comprising them, as well as their use as a medicine.

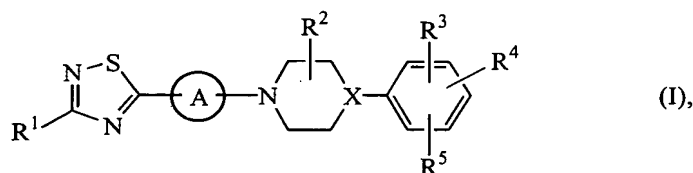
10 Angiogenesis, *i.e.* the formation of new vessels by endothelial cells, plays an important role in a variety of physiologic and pathophysiologic processes. The development of a vascular supply is essential for the growth, maturation and maintenance of normal tissues. It is also required for wound healing. Angiogenesis is critical for solid tumor growth and metastasis and is involved in a variety of other pathological conditions such as neovascular glaucoma, diabetic retinopathy, psoriasis and rheumatoid arthritis. These pathological states are characterized by augmented angiogenesis during which
15 normally quiescent endothelial cells become activated, degrade extracellular matrix barriers, proliferate, and migrate to form new vessels. To control these angiogenesis dependent disorders, compounds with angiogenesis inhibitory properties would be very useful.

20 Several compounds inhibiting angiogenesis, also called angiostatics, angio-inhibitors or angiogenic antagonists, are disclosed in the art. For instance hydrocortisone is a well known angiogenesis inhibitor (Folkman et al., Science 230:1375, 1985' "A new class of steroids inhibits angiogenesis in the presence of heparin or a heparin fragment"; Folkman et al., Science 221:719, 1983, "Angiogenesis inhibition and tumor regression
25 caused by heparin or a heparin fragment in the presence of cortisone").

EP-0,398,427, publised on November 22 1990, discloses antirhinoviral pyridazinames, and in EP-0,435,381, published on July 3 1991, pyridazinamines are described having antipicornaviral activity. EP-0,429,344, published on May 29 1991, discloses
30 aminopyridazine derivatives as cholinergic agonists.

The compounds of the present invention differ from the prior art compounds by the fact that they are invariably substituted with a thiadiazolyl moiety and particularly by the fact that unexpectedly these compounds have angiogenesis inhibiting properties.
35

This invention concerns compounds of formula



- 5 the *N*-oxide forms, the pharmaceutically acceptable acid addition salts and stereochemically isomeric forms thereof, wherein

X is CH or N;

- 10 R¹ is hydrogen, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, amino, mono- or di(C₁₋₆alkyl)amino, Ar¹, Ar¹NH-, C₃₋₆cycloalkyl, hydroxymethyl or benzyloxymethyl;

R² is hydrogen, C₁₋₆alkyl, amino, aminocarbonyl, mono- or di(C₁₋₆alkyl)amino, C₁₋₆alkyloxycarbonyl, C₁₋₆alkylcarbonylamino, hydroxy or C₁₋₆alkyloxy;

- 15 R³, R⁴ and R⁵ are each independently selected from hydrogen, halo, C₁₋₆alkyl, C₁₋₆alkyloxy, trifluoromethyl, nitro, amino, cyano, azido, C₁₋₆alkyloxyC₁₋₆alkyl, C₁₋₆alkylthio, C₁₋₆alkyloxycarbonyl or Het¹;

—(A)— is Ar², Ar²CH₂- or Het²;

Ar¹ is phenyl; phenyl substituted with 1, 2 or 3 substituents each independently selected from halo, C₁₋₆alkyl, C₁₋₆alkyloxy, trihalomethyl, amino or nitro;

- 20 Ar² is phenyl; phenyl substituted with 1, 2 or 3 substituents each independently selected from halo, C₁₋₆alkyl, C₁₋₆alkyloxy, trihalomethyl, amino or nitro;

Het¹ is a monocyclic heterocycle selected from oxazolyl, isoxazolyl, oxadiazolyl, thiazolyl, isothiazolyl, thiadiazolyl or oxazolinyl; and each monocyclic heterocycle may optionally be substituted on a carbon atom with C₁₋₄alkyl; and

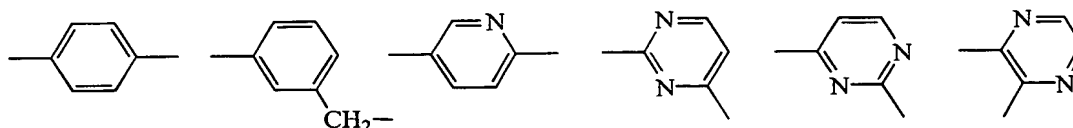
- 25 Het² is a monocyclic heterocycle selected from furanyl, thiofuranyl, oxadiazolyl, thiadiazolyl, pyridinyl, pyrimidinyl or pyrazinyl; and each monocyclic heterocycle may optionally be substituted on a carbon atom with 1 or 2 substituents each independently selected from halo, C₁₋₄alkyl, C₁₋₄alkyloxy, nitro or trifluoromethyl.

- 30 As used in the foregoing definitions and hereinafter, halo is generic to fluoro, chloro, bromo and iodo; C₁₋₄alkyl defines straight and branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as, for example, methyl, ethyl, propyl, butyl, 1-methylethyl, 2-methylpropyl and the like; C₁₋₆alkyl is meant to include

C₁₋₄alkyl and the higher homologues thereof having 5 to 6 carbon atoms such as, for example, pentyl, 2-methylbutyl, hexyl, 2-methylpentyl and the like.

Examples of the $\text{---}(\text{A})\text{---}$ moiety are

5



Wherever $\text{---}(\text{A})\text{---}$ is a radical of formula Ar^2CH_2 , the CH_2 moiety of said radical preferably is connected to the nitrogen atom of the piperidiny moiety when X is CH, or piperazinyl moiety when X is nitrogen.

10

The pharmaceutically acceptable acid addition salts as mentioned hereinabove are meant to comprise the therapeutically active non-toxic acid addition salt forms which the compounds of formula (I) are able to form. The compounds of formula (I) which have basic properties can be converted in their pharmaceutically acceptable acid addition salts by treating said base form with an appropriate acid. Appropriate acids comprise, for example, inorganic acids such as hydrohalic acids, e.g. hydrochloric or hydrobromic acid; sulfuric; nitric; phosphoric and the like acids; or organic acids such as, for example, acetic, propanoic, hydroxyacetic, lactic, pyruvic, oxalic, malonic, succinic (*i.e.* butanedioic acid), maleic, fumaric, malic, tartaric, citric, methanesulfonic, ethanesulfonic, benzenesulfonic, *p*-toluenesulfonic, cyclamic, salicylic, *p*-amino-salicylic, pamoic and the like acids.

15
20

The term acid addition salts also comprises the hydrates and the solvent addition forms which the compounds of formula (I) are able to form. Examples of such forms are e.g. hydrates, alcoholates and the like.

25

The term stereochemically isomeric forms of compounds of formula (I), as used hereinbefore, defines all possible compounds made up of the same atoms bonded by the same sequence of bonds but having different three-dimensional structures which are not interchangeable, which the compounds of formula (I) may possess. Unless otherwise mentioned or indicated, the chemical designation of a compound encompasses the mixture of all possible stereochemically isomeric forms which said compound may possess. Said mixture may contain all diastereomers and/or enantiomers of the basic molecular structure of said compound. All stereochemically isomeric forms of the

30
35

compounds of formula (I) both in pure form or in admixture with each other are intended to be embraced within the scope of the present invention.

5 Some of the compounds of formula (I) may also exist in their tautomeric forms. Such forms although not explicitly indicated in the above formula are intended to be included within the scope of the present invention.

10 The *N*-oxide forms of the compounds of formula (I) are meant to comprise those compounds of formula (I) wherein one or several nitrogen atoms are oxidized to the so-called *N*-oxide.

Whenever used hereinafter, the term "compounds of formula (I)" is meant to include also the pharmaceutically acceptable acid addition salts and all stereoisomeric forms.

15 A group of interesting compounds consists of those compounds of formula (I) wherein one or more of the following restrictions apply :

- a) X is N;
- b) R¹ is hydrogen, C₁₋₆alkyl, amino or di(C₁₋₆alkyl)amino;
- c) R² is hydrogen;
- 20 d) R³, R⁴ and R⁵ are each independently selected from hydrogen, halo, C₁₋₆alkyl, C₁₋₆alkyloxy, trifluoromethyl, nitro or C₁₋₆alkyloxycarbonyl.

25 A particular group of compounds are those compounds of formula (I) wherein X is N; R¹ is hydrogen, C₁₋₄alkyl or di(C₁₋₄alkyl)amino; R² is hydrogen; R³, R⁴ and R⁵ are each independently selected from hydrogen, halo, C₁₋₄alkyl, C₁₋₄alkyloxy or trifluoromethyl; and the bivalent radical $\text{---}(\text{A})\text{---}$ is Ar², Ar²CH₂- or Het² wherein Ar² is phenyl, or phenyl substituted with 1 or 2 substituents each independently selected from halo, C₁₋₆alkyl, C₁₋₆alkyloxy, trihalomethyl, amino or nitro; and Het² is thiadiazolyl, pyridinyl, pyrimidinyl or pyrazinyl.

30 A preferred group of compounds are those compounds of formula (I) wherein X is N, R¹ is methyl, R² is hydrogen, R³ and R⁴ are hydrogen and R⁵ is trifluoromethyl.

35 A more preferred group of compounds are those preferred compounds wherein R⁵ is trifluoromethyl situated on the 3-position.

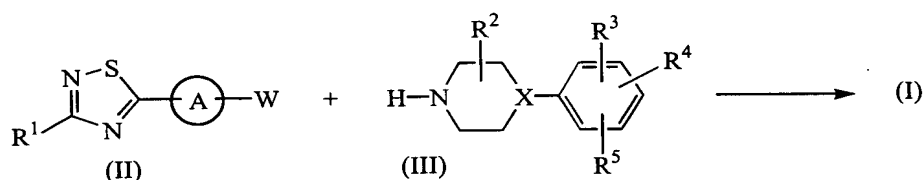
Most preferred are :

1-[4-(3-methyl-1,2,4-thiadiazol-5-yl)phenyl]-4-[3-(trifluoromethyl)phenyl]piperazine,
and

- 1-[5-(3-methyl-1,2,4-thiadiazol-5-yl)-2-pyridinyl]-4-[3-(trifluoromethyl)-phenyl]-
5 piperazine, and the pharmaceutically acceptable acid addition salts, the stereoisomeric
forms, or the *N*-oxides thereof.

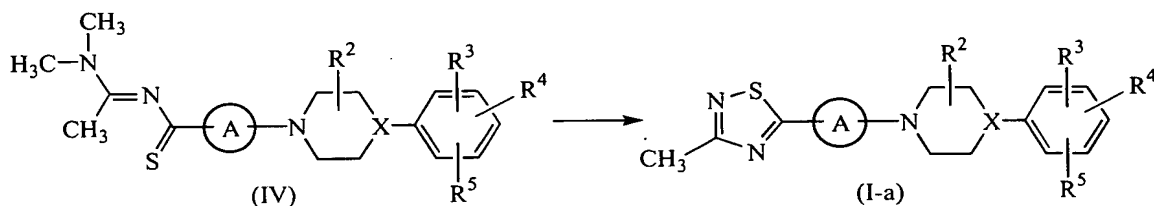
The compounds of the present invention can generally be prepared by reacting an
intermediate of formula (II) with an intermediate of formula (III).

10



- In the foregoing and following reaction schemes W represents an appropriate reactive
leaving group such as, for example, halo, e.g. fluoro, chloro, bromo, iodo, or in some
15 instances W may also be a sulfonyloxy group, e.g. methanesulfonyloxy, benzene-
sulfonyloxy, trifluoromethanesulfonyloxy and the like reactive leaving groups. Said
reaction is performed following art-known procedures such as for instance stirring both
reactants together in a reaction-inert solvent, e.g. *N,N*-dimethylformamide, acetonitrile,
methyl isobutylketone and the like, preferably in the presence of a base, e.g. sodium
20 hydrogen carbonate, sodium carbonate or triethylamine. The reaction may conveniently
be carried out at a temperature ranging between room temperature and the reflux
temperature of the reaction mixture.

- The compounds of formula (I) wherein R^1 is CH_3 , said compounds being represented
25 by formula (I-a) can also be prepared by treating an intermediate of formula (IV) with
hydroxylamino-O-sulfonic acid in a reaction-inert solvent such as, e.g. methanol or
ethanol, in the presence of a base such as, e.g. pyridine.



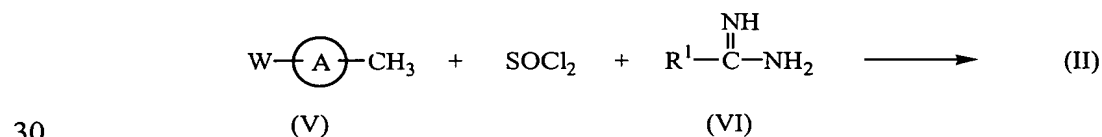
30

The compounds of formula (I) may further be prepared by converting compounds of formula (I) into each other according to art-known group transformation reactions.

5 { The compounds of formula (I) may also be converted to the corresponding *N*-oxide forms following art-known procedures for converting a trivalent nitrogen into its *N*-oxide form. Said *N*-oxidation reaction may generally be carried out by reacting the starting material of formula (I) with an appropriate organic or inorganic peroxide. Appropriate inorganic peroxides comprise, for example, hydrogen peroxide, alkali metal or earth alkaline metal peroxides, e.g. sodium peroxide, potassium peroxide; 10 appropriate organic peroxides may comprise peroxy acids such as, for example, benzenecarboperoxoic acid or halo substituted benzenecarboperoxoic acid, e.g. 3-chlorobenzenecarboperoxoic acid, peroxyalkanoic acids, e.g. peroxyacetic acid, alkylhydroperoxides, e.g. *tert*-butyl hydroperoxide. Suitable solvents are, for example, water, lower alkanols, e.g. ethanol and the like, hydrocarbons, e.g. toluene, ketones, e.g. 15 2-butanone, halogenated hydrocarbons, e.g. dichloromethane, and mixtures of such solvents.

The starting materials and some of the intermediates are known compounds and are commercially available or may be prepared according to conventional reaction 20 procedures generally known in the art. For instance, some intermediates of formula (II), such as 5-(4-fluorophenyl)-3-methyl-1,2,4-thiadiazole, have been described by Yang-i Lin et al. in *J. Org. Chem.*, 45(19), p. 3750 - 3753 (1980), and some intermediates of formula (III), such as 1-[3-(trifluoromethyl)phenyl]-piperazine, are commercially available.

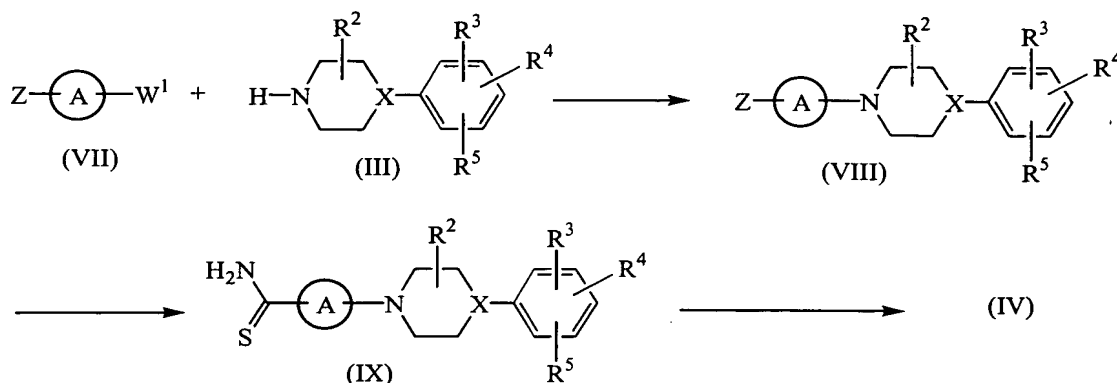
25 Intermediates of formula (II) may be prepared by reacting compounds of formula (V), wherein W is an appropriate leaving group as defined above, with an intermediate of formula (VI), optionally added as its acid addition salt.



Intermediates of formula (IV) can be prepared as outlined in scheme I.

-7-

Scheme I



In scheme I, an intermediate of formula (VII), wherein W¹ is an appropriate leaving group such as halo or sulfonyloxy and Z is cyano or aminocarbonyl, is reacted with an intermediate of formula (III) under art-known reaction procedures as described hereinabove for the synthesis of compounds of formula (I). The resulting intermediates of formula (VIII) are treated with Lawesson's reagent in a suitable solvent such as, for example, toluene or pyridine, or are treated with H₂S in a suitable solvent such as, e.g. *N,N*-dimethylformamide, optionally in the presence of triethylamine. Subsequently, intermediates of formula (IX) are treated with *N,N*-dimethylacetamide dimethyl acetal in a reaction-inert solvent such as, e.g. toluene or dichloromethane, thereby yielding intermediates of formula (IV).

Compounds of formula (I) and some of the intermediates may have one or more stereogenic centers in their structure, present in a R or a S configuration. For instance, R¹, R², R³, R⁴ or R⁵ can be a C₁₋₆alkyl having a stereogenic center.

The compounds of formula (I) as prepared in the hereinabove described processes may be synthesized in the form of racemic mixtures of enantiomers which can be separated from one another following art-known resolution procedures. The racemic compounds of formula (I) may be converted into the corresponding diastereomeric salt forms by reaction with a suitable chiral acid. Said diastereomeric salt forms are subsequently separated, for example, by selective or fractional crystallization and the enantiomers are liberated therefrom by alkali. An alternative manner of separating the enantiomeric forms of the compounds of formula (I) involves liquid chromatography using a chiral stationary phase. Suitable chiral stationary phases are, for example, polysaccharides, in particular cellulose or amylose derivatives. Commercially available polysaccharide

based chiral stationary phases are ChiralCel CA, OA, OB, OC, OD, OF, OG, OJ and OK, and Chiralpak AD, AS, OP(+) and OT(+). Appropriate eluents or mobile phases for use in combination with said polysaccharide chiral stationary phases are hexane and the like, modified with an alcohol such as ethanol, isopropanol and the like. Said pure stereochemically isomeric forms may also be derived from the corresponding pure stereochemically isomeric forms of the appropriate starting materials, provided that the reaction occurs stereospecifically. Preferably if a specific stereoisomer is desired, said compound will be synthesized by stereospecific methods of preparation. These methods will advantageously employ enantiomerically pure starting materials.

The compounds of formula (I) have valuable pharmacological properties in that they inhibit angiogenesis, both *in vivo* and *in vitro*.

In view of their pharmacological activity, the compounds of formula (I), their pharmaceutically acceptable acid addition salts, stereochemically isomeric forms, or *N*-oxide forms thereof, are inhibitors of angiogenesis. Therefore, angiogenesis inhibitors are useful to control or treat angiogenesis dependent disorders such as, e.g. ocular neovascular diseases, neovascular glaucoma, diabetic retinopathy, retrolental fibroplasia, hemangiomas, angiofibromas, psoriasis, osteoarthritis and rheumatoid arthritis. Also, angiogenesis inhibitors are useful to control solid tumor growth, such as, e.g. breast, prostate, melanoma, renal, colon, cervical cancer and the like; and metastasis.

Hence, the present invention discloses the compounds of formula (I) for use as a medicine as well as the use of these compounds of formula (I) for the manufacture of a medicament for treating angiogenesis dependent disorders.

In view of the usefulness of the subject compounds in the treatment or prevention of angiogenesis dependent disorders, the present invention provides a method of treating warm-blooded animals suffering from such disorders, said method comprising the systemic administration of a therapeutic effective amount of a compound of formula (I), a *N*-oxide or a pharmaceutically acceptable acid addition salt thereof.

In view of their useful pharmacological properties, the subject compounds may be formulated into various pharmaceutical forms for administration purposes.

To prepare the pharmaceutical compositions of this invention, an effective amount of the particular compound, in base or acid addition salt form, as the active ingredient is

combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are desirably in unitary dosage form suitable, preferably, for administration orally, rectally or by parenteral injection.

5 For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed, such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs and solutions; or solid carriers such as starches, sugars, kaolin, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules and
10 tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit form, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example, to aid solubility, may be included. Injectable solutions, for example, may be prepared in
15 which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined
20 with suitable additives of any nature in minor proportions, which additives do not cause a significant deleterious effect to the skin. Said additives may facilitate the administration to the skin and/or may be helpful for preparing the desired compositions. These compositions may be administered in various ways, e.g., as a transdermal patch, as a spot-on, as an ointment. Acid addition salts of (I) due to their increased water
25 solubility over the corresponding base form, are obviously more suitable in the preparation of aqueous compositions.

It is especially advantageous to formulate the aforementioned pharmaceutical compositions in dosage unit form for ease of administration and uniformity of dosage. Dosage unit form as used in the specification and claims herein refers to physically
30 discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such dosage unit forms are tablets (including scored or coated tablets), capsules, pills, powder packets, wafers, injectable solutions or suspensions, teaspoonfuls, tablespoonfuls and the like, and
35 segregated multiples thereof.

For oral administration, the pharmaceutical compositions may take the form of solid dose forms, for example, tablets (both swallowable-only and chewable forms), capsules or gelcaps, prepared by conventional means with pharmaceutically acceptable excipients such as binding agents (e.g. pregelatinised maize starch, polyvinylpyrrolidone or hydroxypropyl methylcellulose); fillers (e.g. lactose, microcrystalline cellulose or calcium phosphate); lubricants e.g. magnesium stearate, talc or silica); disintegrants (e.g. potato starch or sodium starch glycollate); or wetting agents (e.g. sodium lauryl sulphate). The tablets may be coated by methods well known in the art.

10

Liquid preparations for oral administration may take the form of, for example, solutions, syrups or suspensions, or they may be presented as a dry product for constitution with water or other suitable vehicle before use. Such liquid preparations may be prepared by conventional means, optionally with pharmaceutically acceptable additives such as suspending agents (e.g. sorbitol syrup, methylcellulose, hydroxypropyl methylcellulose or hydrogenated edible fats); emulsifying agents (e.g. lecithin or acacia); non-aqueous vehicles (e.g. almond oil, oily esters or ethyl alcohol); and preservatives (e.g. methyl or propyl p-hydroxybenzoates or sorbic acid).

15

Pharmaceutically acceptable sweeteners comprise preferably at least one intense sweetener such as saccharin, sodium or calcium saccharin, aspartame, acesulfame potassium, sodium cyclamate, alitame, a dihydrochalcone sweetener, monellin, stevioside or sucralose (4,1',6'-trichloro-4,1',6'-trideoxygalactosucrose), preferably saccharin, sodium or calcium saccharin, and optionally a bulk sweetener such as sorbitol, mannitol, fructose, sucrose, maltose, isomalt, glucose, hydrogenated glucose syrup, xylitol, caramel or honey.

20

25

Intense sweeteners are conveniently employed in low concentrations. For example, in the case of sodium saccharin, the concentration may range from 0.04% to 0.1% (w/v) based on the total volume of the final formulation, and preferably is about 0.06% in the low-dosage formulations and about 0.08% in the high-dosage ones. The bulk sweetener can effectively be used in larger quantities ranging from about 10% to about 35%, preferably from about 10% to 15% (w/v).

30

The pharmaceutically acceptable flavours which can mask the bitter tasting ingredients in the low-dosage formulations are preferably fruit flavours such as cherry, raspberry, black currant or strawberry flavour. A combination of two flavours may yield very

35

good results. In the high-dosage formulations stronger flavours may be required such as Caramel Chocolate flavour, Mint Cool flavour, Fantasy flavour and the like pharmaceutically acceptable strong flavours. Each flavour may be present in the final composition in a concentration ranging from 0.05% to 1% (w/v). Combinations of said strong flavours are advantageously used. Preferably a flavour is used that does not undergo any change or loss of taste and colour under the acidic conditions of the formulation.

The compounds of the invention may also be formulated as depot preparations. Such long acting formulations may be administered by implantation (for example subcutaneously or intramuscularly) or by intramuscular injection. Thus, for example, the compounds may be formulated with suitable polymeric or hydrophobic materials (for example as an emulsion in an acceptable oil) or ion exchange resins, or as sparingly soluble derivatives, for example as a sparingly soluble salt.

The compounds of the invention may be formulated for parenteral administration by injection, conveniently intravenous, intramuscular or subcutaneous injection, for example by bolus injection or continuous intravenous infusion. Formulations for injection may be presented in unit dosage form e.g. in ampoules or in multidose containers, with an added preservative. The compositions may take such forms as suspensions, solutions or emulsions in oily or aqueous vehicles, and may contain formulatory agents such as isotonicizing, suspending, stabilising and/or dispersing agents. Alternatively, the active ingredient may be in powder form for constitution with a suitable vehicle, e.g. sterile pyrogen-free water before use.

The compounds of the invention may also be formulated in rectal compositions such as suppositories or retention enemas, e.g. containing conventional suppository bases such as cocoa butter or other glycerides.

For intranasal administration the compounds of the invention may be used, for example, as a liquid spray, as a powder or in the form of drops.

Those skilled in the art could easily determine the effective amount from the test results presented hereinafter. In general it is contemplated that an effective amount would be from 0.001 mg/kg to 10 mg/kg body weight, and in particular from 0.01 mg/kg to 1 mg/kg body weight. It may be appropriate to administer the required dose as two, three, four or more sub-doses at appropriate intervals throughout the day. Said sub-doses may

-12-

be formulated as unit dosage forms, for example, containing 0.01 to 500 mg, and in particular 0.1 mg to 200 mg of active ingredient per unit dosage form.

The following examples are provided for purposes of illustration

5 Experimental part

Hereinafter "DMF" means *N,N*-dimethylformamide, "DCM" means dichloromethane, "DIPE" means diisopropylether and "THF" means tetrahydrofuran.

A. Preparation of the intermediates

10 Example A.1

a) A mixture of 2-chloro-4-methylpyrimidinyl (0.07 mol) in thionyl chloride (100 ml) was stirred and refluxed for 16 hours. The solvent was evaporated yielding 2-chloro-4-[dichloro(chlorothio)methyl]pyrimidine (intermediate 1).

15 b) 1-Imino-ethanamine hydrochloride (1:1) (0.08 mol) was added at 0°C to a stirring mixture of intermediate 1 (0.07 mol) in DCM (300 ml). Sodium hydroxide (50%, 20 ml) was added dropwise at 0°C. The mixture was stirred at 5°C for 1 hour. Water (300 ml) and DCM (300 ml) were added. The mixture was separated into its layers. The aqueous layer was washed twice with DCM. The combined organic layer was dried, filtered and the solvent was evaporated. The residue was purified over silica gel
20 on a glass filter (eluent : DCM). Two pure fractions were collected and their solvents were evaporated, yielding 3.5g of 2-chloro-4-(3-methyl-1,2,4-thiadiazol-5-yl)pyrimidine (intermediate 2).

Example A.2

25 a) A mixture of 6-chloro-3-pyridinecarboxamide (0.11 mol), 1-[3-(trifluoromethyl)-phenyl]-piperazine (0.11 mol) and sodium carbonate (0.22 mol) in DMF (300 ml) was stirred at 120°C overnight. The mixture was poured out into ice water (600ml) and stirred for 1 hour. The precipitate was filtered off and dried. Part of this fraction (4 g) was taken up in DCM and an aqueous NaHCO₃ solution. The mixture was separated
30 into its layers. The aqueous layer was extracted three times with DCM. The combined organic layer was dried, filtered and the solvent was evaporated till a small volume. The precipitate was filtered off and dried, yielding 3.2 g of 6-[4-[3-(trifluoromethyl)phenyl]-1-piperazinyl]-3-pyridinecarboxamide (intermediate 3).

35 b) A mixture of intermediate 3 (0.013 mol) and Lawesson's reagents (0.007 mol) in toluene (130 ml) was stirred and refluxed for 2 hours. The mixture was cooled. Water (100 ml) was added. The mixture was stirred for 1 hour and separated into its layers.

The aqueous layer was extracted three times with toluene and once with DCM. The combined organic layer was dried, filtered and the solvent was evaporated, yielding: 7.3g of 6-[4-[3-(trifluoromethyl)phenyl]-1-piperazinyl]-3-pyridinecarbothioamide (intermediate 4).

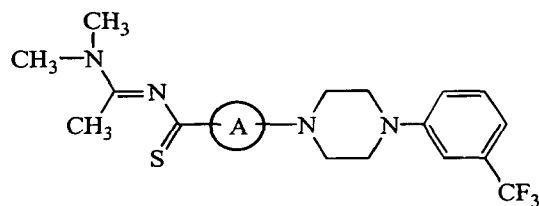
5

Example A.3

A mixture of intermediate 4 (0.013 mol) and 1,1-dimethoxy-*N,N*-dimethyl-ethanamine (0.021 mol) was allowed to stand overnight and then used without further purification, yielding *N*-[1-(dimethylamino)ethylidene]-6-[4-[3-(trifluoromethyl)phenyl]-1-piperazinyl]-3-pyridinecarbothioamide (intermediate 5).

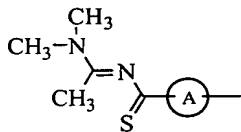
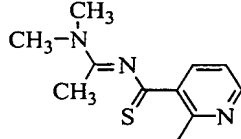
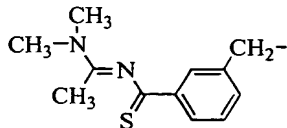
10

Table I.1 lists the intermediates that were prepared according to example A.3.

Table I.1 :

15

Intm. No.	Ex. No.		Physical data
5	A.3		-
6	A.3		mp. 156°C
7	A.3		-
8	A.3		-

Intm. No.	Ex. No.	Chemical Structure	Physical data
			
9	A.3		-
10	A.3		-

B. Preparation of the final compounds

Example B.1

A mixture of 5-(4-fluorophenyl)-3-methyl-1,2,4-thiadiazole (0.012 mol), 1-[3-(trifluoromethyl)phenyl]-piperazine (0.014 mol) and sodium carbonate (0.024 mol) in DMF (10 ml) was stirred at 140°C for 24 hours, then at 150°C for 24 hours, cooled, poured out into ice water (200 ml) and stirred. The precipitate was filtered off, taken up in DCM, dried, filtered and the solvent was evaporated. The residue was crystallized from DIPE. The precipitate was filtered off and dried, yielding 2.5g (52%) of 1-[4-(3-methyl-1,2,4-thiadiazol-5-yl)phenyl]-4-[3-(trifluoromethyl)phenyl]piperazine (compound 2).

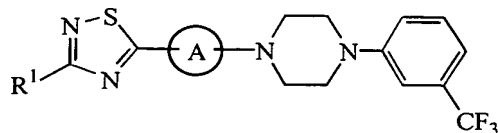
Example B.2

A mixture of hydroxylamine-*O*-sulfonic acid (0.011 mol) in methanol (15 ml) was added at once to a mixture of intermediate 5 (0.01 mol) and pyridine (0.02 mol) in ethanol (40 ml). The mixture was stirred at room temperature for 90 minutes. The solvent was evaporated. The residue was dissolved in DCM, washed with water and an aqueous NaOH 0.1N solution, dried, filtered and the solvent was evaporated. The residue was taken up in methanol, filtered off and dried. The residue was taken up in acetonitrile (100 ml). The mixture was stirred and boiled until complete dissolution and then allowed to crystallize out. The precipitate was filtered off and dried, yielding 1.4 g (35%) of 1-[5-(3-methyl-1,2,4-thiadiazol-5-yl)-2-pyridinyl]-4-[3-(trifluoromethyl)phenyl]piperazine (compound 8).

Table F.1 lists the compounds that were prepared according to one of the above examples and table F.2 lists both the experimental (column heading "exp.") and

theoretical (column heading "theor.") elemental analysis values for carbon, hydrogen and nitrogen of the compounds as prepared in the experimental part hereinabove.

Table F.1 :



5

Co. No.	Ex. No.		Physical data
1	B.1		mp. 184.8°C
2	B.1		-
3	B.2		-
4	B.2		.HCl (1:1)
5	B.2		.HCl (1:1)
6	B.2		-
7	B.2		-
8	B.1		-
9	B.1		mp. 140°C
10	B.2		mp. 140°C

Table F.2 :

Comp. No.	Carbon		Hydrogen		Nitrogen	
	Exp.	Theor.	Exp.	Theor.	Exp.	Theor.
2	59.39	59.31	4.73	4.68	13.85	13.88
3	59.39	59.32	4.73	4.67	13.85	13.92
5	55.44	55.29	4.87	4.96	12.32	12.39
6	56.29	55.30	4.47	4.33	17.27	17.06
7	56.29	56.24	4.47	4.45	17.27	17.43
8	53.19	52.21	4.22	4.07	20.68	20.33

C. Pharmacological examples5 Example C.1

Angiogenesis inhibitory activity was measured *in vitro* using the rat aortic ring model of angiogenesis as described by Nicosia, R.F. and Ottinetti in "Laboratory Investigation", vol. 63, p. 115, 1990. The ability of compounds to inhibit microvessel formation was compared to vehicle-treated control rings. Quantitation (microvessel area) following eight days in culture was performed using an image analysis system, consisting of a light microscope, a CCD camera and an automated, custom-designed image analysis program as described by Nissanov, J., Tuman, R.W., Gruver, L.M., and Fortunato, J.M. in "Laboratory Investigation", vol 73 (#5), p. 734, 1995. Compounds were tested at several concentrations for determination of inhibitory potency (IC₅₀'s).

15 Compounds 1, 2 and 6 have an IC₅₀ value lower than 10 nM.